ALKALOIDS OF ALSTONIA SPHAEROCAPITATA*

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Abstract—Twenty alkaloids have been isolated from the leaves, fruit and stem-bark of Alstonia sphaerocapitata from New Caledonia They were vincamedine, 10-methoxyvincamedine, Z-isositsirikine, akuammicine, quaternoline, 11-methoxyakuammicine, tubotaiwine, 10-methoxyvincamedine N(4)-oxide, cabucraline, cathafoline, caberoline, vincondine, quaternoxine, nor C-fluorocurarine, desoxycabufiline, nordesoxycabufiline and three alkaloids of unknown structure

INTRODUCTION

Alstonia sphaerocapitata Boit is an 8-15 m high tree of New Caledonia. It is a new species, described by Boiteau in 1977 [1]. As part of our chemotaxonomic work on the Alstonia from New Caledonia, we herein describe our results on its alkaloid content.

RESULTS AND DISCUSSION

Extractions were conducted in the usual fashion and the yield of alkaloids was 22 g/kg in the leaves, 2 g/kg in the fruit and 6 6 g/kg in the stem-bark. The alkaloid mixtures (AM) were separated by a combination of medium pressure LC and prep TLC

Twelve identified alkaloids were isolated from the leaves They were, in order of increasing polarity vincamedine 1 (1% of AM), 10-methoxyvincamedine 2 (2%), Z-isositsirikine 3 (0 02 %), akuammicine 4 (3 %), quaternoline 5 (05%), 11-methoxyakuammicine 6 (4%), tubotaiwine 7 (0.5%), 10-methoxy vincamedine N(4)-oxide 8 (2%), cabucraline 9 (8%), cathafoline 10 (8%), caberoline 11 (0.5%) and desoxycabufiline 12 (0.5%) Alkaloid 10 is also known as alkaloid X from Catharanthus roseus [2] and although not definitively proved, quaternoline 5 seems to be identical to raucubaine [3] Compounds 1, 4-7, 9-12 are known compounds identified by direct comparison with reference samples. The major alkaloids of the leaves are the pairs cathafoline 10/cabucraline 9, quaternoline 5/caberoline 11, which only differ by a methoxyl substituent on the aromatic ring, separation of 9 and 10 is tedious and can only be achieved by multiple migration TLC

Vincamedine 1 (vincamajine acetate), also isolated from A deplanchei [4], is accompanied in A sphaerocapitata by compound 2, which possesses very similar spectral properties. Their ¹H NMR spectra are superimposable except for the aromatic area and for a three proton singlet at $\delta 37$, which is absent in the spectrum of 1. The mass spectrum of

Compound 8 displays spectral properties (UV, mass spectrum, IR) similar to those of 2, slight differences are found in the 1 H NMR spectra, in the $\delta 40-45$ region,

² shows an $[M]^+$ at m/z 438 which analysed for $C_{25}H_{30}N_2O_5$ Fragments pertaining to the indole part of the molecule are shifted 30 mu above the corresponding fragments of 1 These data led to the conclusion that 2 is a methoxylated vincamedine Location of the methoxyl group on the indole ring is secured by comparison of the ¹³C NMR spectra of 1 and 2 (Table 1) As anticipated, these spectra are superimposable (± 0.1 ppm) except for those carbons related to the aromatic ring We propose C-10 as the substitution position in 2 to account for the shifts of the aromatic carbons and especially that of C-12 ($\delta 1110$) Ample precedent exists in the literature for such a use of ¹³C shifts in the determination of a methoxyl substituent on a dihydroindole ring [5, 6]

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Table 1 13 C NMR spectral data of 1 and 2 (CDCl₃, δ , TMS, attributions according to ref [5])

Position	1	2	Position	1	2
C-2	75 6	75 6	C-16	59 0	59 1
C-3	53 2	53 3	C-17	750	75 1
C-5	617	61 7	C-18	127	127
C-6	36 5	366	C-19	1167	1167
C-7	56 2	56 2	C-20	1369	1369
C-8	129 1	1306	C-21	556	55 7
C-9	123 3	109 7	N-Me	34 2	349
C-10	1189	148 7	C-10-OMe		559
C-11	128 7	113 1	COOMe	1723	172 3
C-12	109 4	1110	COOMe	51 5	51 5
C-13	154 5	1537	OCOMe	168 4	168 4
C-14	219	21 9	OCOMe	20 7	208
C-15	30 5	30 5			

which corresponds to the H-C-N protons of the molecule The hypothesis of 8 being the N-oxide of 2 stems from their behaviour on TLC (different mobilities but identical colours with spray reagents), this has been confirmed by the conversion of 2 into 8 using p-nitroperbenzoic acid

The non polar alkaloid 3 is an indole as shown by its UV absorption (227, 283 and 290 nm) Its mass spectrum is reminiscent of that of the sitsirikines. In these compounds, the $[M]^+$ at m/z 354 is accompanied by an intense $[M-1]^+$ ion and by fragments corresponding to the loss of H_2O (m/z 336), CH_2OH (m/z 323), CO_2Me (m/z 295) and CH₂OH-CH-CO₂Me (m/z 251, 100%)Ions at m/z 156, 169, 170 and 184 derive from the β carboline part of the molecule Compound 3 belongs to the isositsirikine series and not to the sitsirikine series, as demonstrated by the ¹H NMR spectrum of 3, in which a quartet at $\delta 5$ 45 and a three proton doublet at 1 73 signify an ethylidene side chain Direct comparison of 3 and of 16R and of 16S-isositsirikines showed the three compounds to be different Recent publications by Husson et al [7] on the reduction products of 4,21-dehydrogeissoschizine and by Cordell et al [8] on the isolation of 16S Zisositsirikine lead us to consider that 3 could be the missing 16R Z-isositsirikine That this was the case was demonstrated by comparison of our data and compounds with those of H P Husson Two unidentified monomers 18 and 19 have also been isolated, they are described in the Experimental

Similar extraction of the fruit gave a mixture in which 11 known alkaloids were separated. In order of increasing polarity, they are vincoridine 13 (0.5% of AM), vincamedine 1 (5%), 10-methoxyvincamedine 2 (5%), quaternoxine 15 (0.5%), akuammicine 4 (11%), 11-methoxyakuammicine 6 (9%), 10-methoxyvincamedine N-oxide 8 (1%), cabucraline 9 (17%), cathafoline 10 (11%) and desoxycabufiline 12 (8%) Among these alkaloids, compounds 1, 2, 4, 6 and 8-12 are also found in the leaves, alkaloids 13-15 were identified by direct comparison with authentic samples

Analogous treatment of the stem-bark was also accomplished and 11 alkaloids were separated. They are, in order of increasing polarity vincamedine 1 (2% of AM), 10-methoxyvincamedine 2 (3%), akuammicine 4 (7%), 11-methoxyakuammicine 6 (85%), nor C-fluorocurarine 16

(1%), 10-methoxyvincamedine N(4)-oxide 8 (1%), cabucraline 9 (estimated 26% of AM), cathafoline 10 (estimated 7%), desoxycabufiline 12 (5%), nor desoxycabufiline 17 (05%) and an unknown dimeric alkaloid 20 (1%) Compound 17 as well as compound 12 have also been found and identified in A plumosa [9, 10]

A sphaerocapitata, as in all other Alstonia from New Caledonia, contains mainly indole alkaloids of type I of the Le Men-Taylor classification. However, it offers a rare combination of a tetracyclic alkaloid (16R Z-isositsirikine 3) and of the corresponding alkaloids possessing a C-16/C-7 bond (9, 10, 11, 13 and 15) or a C-16/C-5 bond (1, 2, 8 and 14), 'rearranged' alkaloids with a C-16/C-2 bond (4, 6 and 16) are also present

As all the New Caledonian species of Alstonia, A sphaerocapitata belongs to the section Dissuraspermum Monachino, same considerations as mentioned in the work on A plumosa are of value here [9]

EXPERIMENTAL

General Mps are uncorr NMR were measured in CDCl₃ solns at 60 MHz or at 400 MHz on a prototype at the Institut d'Electronique Fondamentale (Orsay) Chemical shifts are given in δ -values with TMS as int standard, coupling constants are given in Hz Rotations are measured in CHCl₃ (10 cm cell) Medium pressure LC expts were run at 10 bar Colour reactions (CR) were obtained by spraying plates with a soln of Ce(IV) (NH₄)₂SO₄ Plant material was collected on the East Coast in a lowland rainforest near Poindimie (6 July, 1979) under reference Sevenet-Pusset 1722, an herbarium specimen is kept in the Herbarium of ORSTOM Center in Noumea

Extraction and isolation of alkaloids Finely ground leaves (1 04 kg) were wetted with 50% NH₄OH (624 ml—48 hr) and lixiviated by means of 261 of EtOAc The lixiviate was extracted with 2% H2SO4 and the aq phase made alkaline with NH4OH and thoroughly extracted with CHCl3 The CHCl3 layers were dried (Na₂SO₄) and evapd in vacuo to give 237 g of crude alkaloid mixture (yield 22 7 g/kg) The AM was fractionated on 1 kg silica gel H60 (elution pressure 10 bar, 30 ml fractions), the column was eluted with CHCl₃ (31), CHCl₃-MeOH (49 1, 121, 19 1, 61, 9 1, 21) The tubes were analysed by TLC and pooled according to their composition Vincamedine 1 and unknowns 18 and 19 were in fractions 101-118, 10-methoxy vincamedine 2 (430 mg) was in fractions 119-180 and Z-isositsirikine 3 in fractions 181-190 The content of tubes 225-265 was refractionated by CC to yield akuammicine 4, quaternoline 5 and 11-methoxyakuammicine 6, tubotaiwine 7 and the N(4)-oxide of 10-methoxyvincamedine 8 were found in fr 281-290 The major alkaloids cabucraline 9 (175 g), cathafoline 10 (178 g) and caberoline 11 (015 g) were in fractions 300-360. The dimer desoxy cabufiline 12 was cluted with CHCl3-McOH (9 1) in fr 361-427

In the same manner 1 08 g of crude AM was obtained from 540 g of dried fruit (yield 2 g/kg) The AM was fractionated on 250 g silica gel H60, elution solvents were CH_2Cl_2 (3 61), CH_2Cl_2 —MeOH (99 1, 3 61, 49 1, 2 41, 19 1, 4 21, 9 1, 3 31, 4 1, 1 81) and MeOH (1 51) Vincoridine 13 (10 mg) was in fractions 81–120, 1 and 2 in fr 196–240, quebrachidine 14 (54 mg) in fr 280–340 quaternoxine 15 (14 mg) in fr 321–343, 4, 6 and 8 in fr 344–434, 9 and 10 in fr 425–570 while 12 was in fr 588–680

Dried stem bark (3 5 kg) yielded 23 g of crude AM which was also separated on 1 kg silica gel Alkaloids 1 and 2 were eluted with CH₂Cl₂-MeOH (49 1) in fr 161-219, 4 in fr 271-470, fr 471-571 yielded 6, 8 and nor C fluorocurarine 16 and 9 and 10 were in fr 572-913 The dimers 12 and 17 were in fr 914-1057 and the unknown 20 in fr 1057-1127

Alkaloids 1, 4–7 and 9–17, available from other studies in the Reims laboratory, were identified by direct comparison as well as by examination of their spectral properties

Description of new alkaloids 10-methoxyvincamedine 2 $C_{25}H_{30}N_2O_5$, (CR pink), $[\alpha]_D = -9^\circ$ (c 1), UV λ_{max}^{MeOH} nm 247, 313, IR $v_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 1750, 1735, 1610, MS (rel int) m/z 438 (100), 423, 379, 264, 222, 200, 190, 187, 174, ¹H NMR (60 MHz) δ 5 68 (s, 1H), 5 25 (q, J = 7 Hz, 1H), 3 7 (s, 3H), 3 6 (s, 3H), 2 65 (s, 3H), 185 (s, 3H), 155 (d, J = 7 Hz, 3H) Unknown 18 (CR yellow), UV \(\lambda_{\text{max}}^{\text{MeOH}} \) nm 233, 280, \(\lambda_{\text{max}}^{\text{MeOH-H}^+} \) nm 225, 258 (sh) 290, IR $v_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3460, 1770, 1635, MS (rel int) m/z 366 (100), 336, 308, 265, 229 (85), 200 (98), 199, 182, 170, 158, ¹H NMR (400 MHz) δ 7 2 (d, J = 2 Hz, 1H), 7 15 (d, J = 8 Hz, 1H), 68 (dd, J = 2 Hz and 8 Hz, 1H), 46 (m, 2H), 385 (s, 3H), 31(s, 3H) Unknown 19 (CR orange), UV λ_{max}^{MeOH} nm 233, 283, IR $v_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 1760, 1635, MS (rel int) m/z 336 (100), 306, 229 (22), 200 (40), 183, 170, 108, 1 H NMR (60 MHz) δ 7 2–6 7 (m, 3H), 38 (s, 3H), 11 (t, J = 7 Hz, 3H) 10-Methoxyvincamedine N(4)-oxide 8 $C_{25}H_{30}N_2O_6$ (CR red), $[\alpha]_D = -2^{\circ}$ (c1), UV λ_{max}^{MeOH} nm 212, 248, 305, IR $\nu_{max}^{CHCl_3}$ cm⁻¹ 1740, MS (rel int) m/z 454, 438 (100), 379, 264, 222, 212, 200, 190, 187, 174, ¹H NMR (60 MHz) $\delta 5$ 65 (s, 1H), 54 (q, J = 7 Hz, 1H), 37 (s, 3H), 3 65 (s, 3H), 2 6 (s, 3H), 1 85 (s, 3H), 1 55 (d, J = 7 Hz, 3H) Z-Isositsirikine 3 C₂₁H₂₆N₂O₃, mp 183°, CR pale yellow green, $[\alpha]_D = -32^\circ$ (c 1), UV λ_{max}^{MeOH} nm 227, 283, 290, IR $v_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3400, 1710, 1450, 1380, MS (rel int) m/z 354 (100), 353, 339, 337, 323, 295, 251, 249, 237, 184, 170, 169, 156, ¹H NMR (60 MHz) δ 8 6 (s, 1H), 7 5–6 9 (m, 4H), 5 45 (q, J = 6 Hz, 1H, 3.7 (s, 3H), 1.7 (d, J = 6 Hz, 3H) Unknown dimer 20(CR purple), $[\alpha]_D = -32^\circ$ (c 1), UV λ_{max}^{MeOH} nm 218, 231 (sh), 256 (sh), 288, 294, IR $v_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹ 3360, 1730, 1670, 1620, MS (rel int) m/z 732 (100), 702, 701, 656, 629, 381, 366, 365 (78), 350, 335, 194, ¹H NMR (400 MHz) δ 7 2 (d, J = 7 Hz, 1H), 7 1 (d, J $= 7 \text{ Hz}, 1 \text{H}, 70 (t, J = 7 \text{ Hz}, 1 \text{H}), 685 (t, J = 7 \text{ Hz}, 1 \text{H}), 62 (s, J = 7 \text{ Hz}, 1 \text{H}), 62 (s, J = 7 \text{ Hz}, 1 \text{H}), 62 (s, J = 7 \text{ Hz}, 1 \text{H}), 62 (s, J = 7 \text{ Hz}, 1 \text{H}), 62 (s, J = 7 \text{ Hz}, 1 \text{H}), 62 (s, J = 7 \text{ Hz}, 1 \text{H}), 62 (s, J = 7 \text{ Hz}, 1 \text{H}), 62 (s, J = 7 \text{ Hz}, 1 \text{Hz}), 62 (s, J = 7 \text{ Hz}, 1 \text{Hz}), 62 (s, J = 7 \text{ Hz}, 1 \text{Hz}), 62 (s, J = 7 \text{ Hz}, 1 \text{Hz}), 62 (s, J = 7 \text{ Hz}, 1 \text{Hz}), 62 (s, J = 7 \text{ Hz}, 1 \text{Hz}), 62 (s, J = 7 \text{ Hz}, 1 \text{Hz}), 62 (s, J = 7 \text{ Hz}, 1 \text{Hz}), 63 (s, J = 7 \text{ Hz}, 1 \text{Hz}), 64 (s, J = 7 \text{ H$

1H), 6 18 (s, 1H), 5 28 (dq, J = 7 Hz, 2H), 4 85 (s, 1H), 4 2 (d, J = 8 Hz, 2H), 3 9 (s, 3H), 3 6 (s, 3H), 3 0 (s, 3H), 2 8 (s, 3H), 2 63 (s, 3H), 1 6 (d, J = 7 Hz, 3H), 1 25 (d, J = 7 Hz, 3H) Oxidation of 10-methoxyvincamedine 2 10-Methoxyvincamedine (20 mg) was dissolved in 2 ml CH₂Cl₂ to which 15 mg of p-nitroperbenzoic acid was added After 30 min to the soln was added 1 ml 2 N NaOH, usual treatment of the organic layer yielded 15 mg of 10-methoxyvincamedine N-oxide identical to 9 (TLC, UV, IR, MS)

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